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**POSSIBILIDADES DE CONTROLE MICROBIANO NA ESTOCAGEM SIMULADA DE
BIODIESEL E MISTURA B20**

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**POSSIBILIDADES DE CONTROLE MICROBIANO NA ESTOCAGEM SIMULADA
DE BIODIESEL E MISTURA B20**

Tese apresentada ao Programa de Pós-Graduação em Microbiologia Agrícola e do Ambiente do Instituto de Ciências Básicas da Saúde da Universidade Federal do Rio Grande do Sul como requisito parcial para a obtenção do título de doutor em Microbiologia Agrícola e do Ambiente.

Orientadora: Prof^a Dr^a Fátima Menezes Bento

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Siglas

°C – Grau Celsius

ABNT – Associação Brasileira de Normas Técnicas

ANP – Agência Nacional do Petróleo, Biocombustíveis e Gás Natural

ASTM – American Society for Testing and Materials

FTIR – Fourier transform infrared spectroscopy

g – grama

GC – Cromatografia Gasosa

GC MS – Cromatografia Gasosa com Espectrômetro de Massas

Kg – quilograma

M – molar

m³ – Metro Cúbico

mg – miligrama

min – minuto

mL – mililitro

mm – milímetro

PCA – Plate Count Agar

pH – potencial hidrogeniônico

ppm – partes por milhão

TSA – Tryptone Soy Agar

UV – ultravioleta

µL – microlitro

µm – micrômetro

POSSIBILIDADES DE CONTROLE MICROBIANO NA ESTOCAGEM SIMULADA DE BIODIESEL E MISTURA B20

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RESUMO

A mistura de biodiesel ao óleo diesel no Brasil trouxe vantagens ambientais e estratégicas para a matriz energética mundial. No entanto, a presença de água e micro-organismos deteriorativos no armazenamento causa formação de borra biológica e alterações do combustível, sendo recomendada a drenagem diária da água e adição de preservantes. O objetivo deste trabalho foi avaliar a atividade antimicrobiana residual de compostos organoestânicos, utilizados na síntese de biodiesel e dos antifúngicos clínicos 8-Hidroxiquinolina (8HQ) e Clotrimazol (CLQ) - comparando-os com biocidas comerciais. Foram utilizados os microrganismos deteriorativos *Aureobasidium pullulans*, *Pseudallescheria boydii*, *Penicillium citrinum*, *Pseudomonas aeruginosa*, *Bacillus pumilus* e um inóculo segundo norma E1256 (ASTM, 2018) (inóculo ASTM). Foram determinadas as concentrações mínima inibitória (CMI) e biocida (CMB) para estanho IV, CLQ, 8HQ e quatro biocidas comerciais (A, B, C e D). O efeito da depleção de micronutrientes (Ca^{++} , Fe^{++} , Mg^{++}) e da presença de EDTA foi comparado com o dos antifúngicos. O efeito sinérgico entre CLQ e o biocida C foi avaliado com o inóculo ASTM. Os testes foram feitos em placas de 96 poços ou frascos de 20 mL, a 30 °C por até 28 dias, quando o pH foi medido e a produção de biomassa definida por contagem de Unidades Formadoras de Colônias ou por método gravimétrico. O combustível foi avaliado por análise de Infravermelho (FTIR) e cromatografia gasosa (GC). *B. pumilus* foi inibido pelo estanho, assim como *A. pullulans*. Os produtos A e B não foram eficazes a 2000 ppm. EDTA a 1000 ppm inibiu o inóculo ASTM e estimulou *P. boydii*. A análise de FTIR indicou degradação do combustível, e GC não indicou alteração do teor de ésteres. Em microcosmos, MIC x 8 não ocasionou efeito biocida ou biostático de CLQ ou 8HQ. Uma mistura de morfolinas, apresentou melhor inibição e efeito sinérgico junto a CLQ sobre o inóculo ASTM.

POSSIBILITIES OF MICROBIAL CONTROL IN BIODIESEL AND B20 BLEND SIMULATED STORAGE

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ABSTRACT

The blending of biodiesel with diesel fuel in Brazil has brought environmental and strategic advantages to the global energy matrix. However, the presence of water and deteriorating microorganisms during storage leads to the formation of biological sludge and fuel alterations, making it advisable to drain water daily and add biocides. The aim of this study was to evaluate the residual antimicrobial activity of organotin compounds used in the synthesis of biodiesel, and of the clinical antifungals 8-Hydroxyquinoline (8HQ) and Clioquinol (CLQ), comparing them with commercial biocides. Deteriogenic microorganisms, including *Aureobasidium pullulans*, *Pseudallescheria boydii*, *Penicillium citrinum*, *Pseudomonas aeruginosa*, *Bacillus pumilus*, and an inoculum following ASTM E1259 standard (ASTM inoculum), were used. Minimum inhibitory concentrations (MIC) and biocide minimum bactericidal concentrations (MBC) were determined for tin IV, CLQ, 8HQ, and four commercial biocides (A, B, C, and D). The effects of micronutrient depletion (Ca⁺⁺, Fe⁺⁺, Mg⁺⁺) and the presence of EDTA were compared to those of the antifungals. The synergistic effect between CLQ and biocide C was evaluated using the ASTM inoculum. The tests were performed in 96-well plates or 20 mL vials at 30°C for up to 28 days, during which the pH was measured, and biomass production was determined by counting Colony-Forming Units or using gravimetric methods. The fuel was assessed by Infrared Analysis (FTIR) and gas chromatography (GC). *B. pumilus* was inhibited by tin, as was *A. pullulans*. Products A and B were not effective at 2000 ppm. EDTA at 1000 ppm inhibited the ASTM inoculum and stimulated *P. boydii*. FTIR analysis indicated fuel degradation, while GC did not show any alteration in ester content. In microcosms, MIC x 8 did not induce a biocidal or biostatic effect of CLQ or 8HQ. A mixture of morpholines showed better inhibition and synergistic effect when combined with CLQ on the ASTM inoculum.

1. INTRODUÇÃO

Combustíveis derivados do petróleo, como o óleo diesel, gasolina e o querosene de aviação, têm sido a principal fonte energética utilizada no mundo, porém, a busca por alternativas renováveis e de menor impacto ambiental é uma realidade. Neste sentido, desde 2008, o óleo diesel comercializado no Brasil é classificado como tipos A como o óleo diesel marítimo (sem biodiesel) ou tipo B, óleo diesel rodoviário (com adição de biodiesel). O percentual mínimo de biodiesel adicionado ao óleo diesel em 2023 é de 12% (constituindo a mistura B12) e a máxima de 15% (mistura B15). Com a implementação da Política Nacional de Biocombustíveis (RENOVABIO), existe uma previsão de aumento deste teor, tornando a mistura B15 obrigatória em 2023. No Brasil foram produzidos 6,3 bilhões de litros de biodiesel em 2022, colocando o Brasil como terceiro maior produtor e consumidor de biodiesel no mercado internacional.

Para a liberação e comercialização no mercado, 24 análises são mandatórias para o biodiesel, segundo a Resolução Nº 45 de 2014 (e devidas alterações) da Agência Nacional de Petróleo e Biocombustíveis. Caso o biodiesel não venha a ser comercializado no prazo de 30 dias após a emissão do Certificado de Qualidade pelo fabricante, o mesmo deverá ser submetido a novas análises. Devido à natureza do biocombustível, composto por ésteres de ácidos graxos, o biodiesel possui maior biodegradabilidade do que o diesel. Em caso de contaminação ambiental essa propriedade é considerada uma vantagem, mas é reconhecida como uma vulnerabilidade durante o armazenamento. Na medida que microrganismos estão presentes no combustível, e em situações de alta densidade microbiana e longos períodos de residência do combustível nos tanques pode ocorrer o comprometimento da qualidade final do produto.

Com a introdução de biodiesel na mistura com o diesel rodoviário verificou-se uma maior suscetibilidade à biodeterioração, devido principalmente a natureza da molécula e a higroscopicidade do biodiesel, resultando em maior incorporação de água durante o armazenamento. Como consequência, foi constatado, desde a introdução do biodiesel em misturas (diesel/biodiesel), um aumento na geração de sedimentos de natureza biológica e/ou química, com o comprometimento da qualidade final do produto armazenado. Estudos sobre o potencial de crescimento de isolados fúngicos filamentosos de lodos biológicos têm mostrado que este grupo de microrganismos apresenta capacidade de degradar hidrocarbonetos e ésteres de ácidos graxos, vantagem metabólica relacionada à presença de genes relacionados à degradação e captação e à grande superfície de contato formada pelas hifas. O micélio pode formar emaranhados de biomassa, ocasionando entupimento de filtros e bombas em tanques e caminhões. As consequências da contaminação microbiana durante o armazenamento estão associadas diretamente ao tempo de residência do

combustível e as rotinas de manutenção adotadas, como drenagens frequentes e limpeza dos tanques, como forma de minimizar a presença de água no fundo dos tanques.

Como estratégia de controle químico de microrganismos são utilizados biocidas, compostos químicos (orgânicos e/ou inorgânicos) com liberação de uso em outros países como Estados Unidos. No Brasil, vários estudos foram conduzidos avaliando a efetividade de moléculas naturais e/ou artificiais quanto ao potencial antimicrobiano. Os produtos devem atender a exigências importantes para o uso em combustível, tais como amplo espectro de ação (fungos e bactérias), coeficiente de partição que permita a solubilidade em ambas as fases óleo-água, baixo impacto ambiental e toxicológico. Dentre os compostos comerciais conhecidos com ação antimicrobiana estão as oxazolidinas, isotiazolonas e morfollinas. Compostos organoestânicos também são conhecidos como antimicrobianos, pelo que neste trabalho foi avaliado o efeito residual de um catalisador a base de estanho (síntese de biodiesel) sobre microrganismos deteriogênicos de diesel-biodiesel. Outro grupo de moléculas avaliado foram as quinollinas. A molécula mais conhecida desta classe é o clloquinol (5-cloro-7-iodo-8-hidroxiquinollina), um derivado de 8-Hidroxiquinollina (8HQ). Tanto o CLQ quanto a 8HQ possuem baixa irritabilidade e toxicidade tópicá, bem como alta eficácia. O mecanismo de ação ocorre especialmente sobre a parede fúngica, atuando sobre a formação de hifas e pseudohifas, o que torna a molécula promissora como biocida contra fungos deteriogênicos.

2. HIPÓTESES:

✓ A presença residual de catalisadores organoestânicos utilizados na síntese de biodiesel pode ter atividade antimicrobiana.

✓ O uso de clioquinol ou hidroxiquinolina como aditivo no armazenamento de biodiesel e mistura B20 pode ter potencial como antimicrobiano, sem provocar alterações nas especificações dos combustíveis.

✓ Os antifúngicos clioquinol e/ou 8-hidroxiquinolina podem apresentar maior efetividade do que biocidas comerciais (Isotiazolonas e Morfolinas) na preservação de combustíveis armazenados.

3. OBJETIVOS

3.1 Objetivo Geral

Avaliar os efeitos antimicrobianos do resíduo de um catalisador a base de estanho utilizado na síntese de biodiesel, de moléculas derivadas do clioquinol sobre fungos deteriorogênicos em biodiesel puro e na mistura diesel e biodiesel (B20) e de biocidas comerciais disponíveis no mercado para controle microbiano no armazenamento.

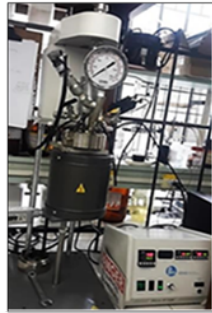
3.2 Objetivos Específicos

- 3.2.1 Avaliar a atividade biocida e/ou bioestática residual de um catalisador organoestânico utilizado na síntese de biodiesel de soja sobre um inóculo não caracterizado segundo norma ASTM 1259-18, sobre os fungos *Aureobasidium pullulans* e *Pseudallescheria boydii* e sobre as bactérias *Pseudomonas aeruginosa* e *Bacillus pumilus*, em microcosmos de 10 mL (estocagem simulada), observando a deterioração abiótica do combustível em comparação com a degradação observada no combustível inoculado.
- 3.2.2 Avaliar a atividade fungicida e/ou fungistática de 8-hidroxiquinolina e clioquinol (determinando Concentração Mínima Inibitória e Concentração Mínima Biocida), em condições ideais e em estocagem simulada (microcosmos) com um inóculo não caracterizado (Norma ASTM 1259-18) e com os fungos *Aureobasidium pullulans*, *Pseudallescheria boydii* e *Penicillium citrinum*.
- 3.2.3 Avaliar atividade fungicida e/ou fungistática de compostos disponíveis comercialmente (determinando Concentração Mínima Inibitória e Concentração Mínima Biocida), em condições ideais e em estocagem simulada (microcosmos) com um inóculo não caracterizado (Norma ASTM 1259-18) e com os fungos *Aureobasidium pullulans*, e *Pseudallescheria boydii*.
- 3.2.4 Avaliar a presença de efeito sinérgico de biocidas comerciais com 8-hidroxiquinolina ou clioquinol.

5. **CAPÍTULO I - Avaliação do efeito antimicrobiano de um catalisador organoestânico sobre microrganismos deteriogênicos e inóculo não-caracterizado em biodiesel (B100)**

O artigo **Characterization of antimicrobial effect of organotin-based catalysts on diesel–biodiesel deteriogetic microorganisms** está redigido em inglês, foi formatado segundo as normas da revista **Environmental Monitoring and Assessment** e publicado em dezembro de 2020 sob o DOI <https://doi.org/10.1007/s10661-020-08744-x>

Graphical Abstract:



Biodiesel Reactor

Catalyst 1:
($C_{32}H_{64}O_4Sn$)
Catalyst 2:
($C_8H_{18}OSn$)

PART 1: Selection of tin catalyst

Catalyst 1
3000 ppm

Catalyst 2
2900 ppm

Viability of ASTM E1259-16 Inoculum after 14 days

Catalyst 1
0 ppm

Catalyst 1
3000 ppm
sterile control

Catalyst 2
0 ppm

Catalyst 2
2900 ppm
sterile control

a

b

PART 2 (Selected catalyst): AFTER 28 days

**Biodiesel
deteriogenic
microorganisms**

FUNGI
Pseudallescheria boydii
Aurobasidium pullulans

BACTERIA
Bacillus pumilus
Pseudomonas aeruginosa

POOL OF MICROORGANISMS
(ASTM E1259-16
Non-characterized
microorganisms)

Interface **Water phase** **Oil phase**

Dry biomass CFU Viability Infrared and chromatographic characterization

Infrared



Characterization of antimicrobial effect of organotin-based catalysts on diesel–biodiesel deteriogenic microorganisms

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Abstract Organotin compounds are applied in several industrial reactions and can present antifungal and antibacterial activities. Incorrect handling and storage practices of biodiesel and diesel–biodiesel blends can lead to microbial development, impacting its final quality. Concerning this problem, this work investigated the antimicrobial action of two organotin catalysts used in biodiesel production with four isolated microorganisms (*Bacillus pumilus*, *Pseudomonas aeruginosa*, *Pseudallescheria boydii*, and *Aureobasidium pullulans*) and a pool of microorganisms (ASTM E1259 standard practice). Samples of soybean biodiesel with different concentrations of dibutyl tin dilaurate (catalyst 1) and di-n-butyl-oxo-stannane (catalyst 2) were prepared and added of mineral medium. The pool of microorganisms was inoculated and incubated at 30 °C and final biomass was weighed after 14 days. Thereafter, soybean biodiesel with catalyst 2 was used. Fungal biomass was weighted, and plate count was used to assess

bacterial growth. Results show that catalysts 1 and 2 presented no inhibitory activity on the pool of microorganisms evaluated. A slight inhibitory activity was observed for *B. pumilus* and *A. pullulans* growth, but not for *P. boydii*, *P. aeruginosa*, or the pool of microorganisms. All experiments exhibited acidification higher than sterile control. Infrared analysis shows less microbiological degradation products in the tin-protected fuel with ASTM inoculum. These results suggest that these tin-based catalysts show no toxic effect on native microbial population and a slight effect on some isolated microbial population in laboratory scale and for the first time shows that these organotin compounds can be employed safely as biodiesel catalyst.

Keywords ASTM · Biocide · Biodeterioration · Microbial control · Preservative · Soybean

Introduction

Brazil has increasingly added biodiesel to fossil diesel, and the National Biofuel policy—RenovaBio—promotes the use of a series of biofuels and at the same time encourages the obtention of Decarbonization Credits—i.e., the difference in emissions between fossil fuels and biofuel (Salina et al., 2020). The use of biodiesel in Brazil started with a compulsory blend of 2% in fossil diesel in 2008 (B2), and today the mandatory blend has at least 12% biodiesel, and until 15% is allowed (RESOLUÇÃO N° 16, DE 29 DE OUTUBRO DE 2018, 2018). The presence of biodiesel in diesel fuel is therefore a reality, which brings environmental and strategic advantages. However, there are still challenges to be overcome, such as the possibility of biodiesel deterioration in its transportation and storage (Passman, 2013;

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Bücker et al., 2014, 2018; Bento et al., 2016; Cazarolli et al., 2020). One of the major problems observed in commercial use of biodiesel is its instability, which can occur during transport and storage under incorrect conditions, compromising the final quality of the biofuel. The presence of free water in the fuels during its transport and storage systems can accelerate both abiotic (oxidation and hydrolysis) and biotic (microbial participation) deterioration processes (Atadashi et al., 2013; Salina et al., 2020). Biodiesel has an amphiphilic nature, which increases the dissolution rate of diesel in water, facilitating its dissolution in water and microbial development (Gupta, 2020). Biodiesel is chemically defined as a mixture of monoalkyl fatty acid esters obtained by transesterification of vegetable or animal oils and fats with methanol or ethanol in the presence of a catalyst (acid or base) (Knothe & Gerpen, 2005; Nautiyal et al., 2020). The presence of water in tank bottoms is the most important factor influencing the extent of microbial growth in fuel tanks and systems. It is well documented that microorganisms (aerobic and anaerobic bacteria; filamentous fungi, yeasts) will find appropriate conditions for growth and they are most metabolically active at the oil–water interface (Fátima Menezes Bento et al., 2016; Cazarolli et al., 2020; Passman, 2013; Zimmer et al., 2017). Good house-keeping is recommended in Brazil by following Brazilian Standards such as ABNT NBR 15512 (ABNT, 2014) and 16732-19 (ABNT, 2019). The imperative for regular draining of water and the addition of chemical compounds exhibiting antimicrobial activity (biocides), to prevent or kill microorganisms are also cited. However, in Brazil, the use of biocides is still under evaluation, as there are still many clarifications to be made concerning their use and disposal in the environment (Luz et al., 2018). A chemical strategy in controlling microbial growth is the use of additives and/or antimicrobial compounds that can act as a fuel preservative at all stages of product handling (Passman, 2013).

Organotin compounds are used in many catalytic activities in industry (Hadi et al., 2019; Zubair et al., 2019) and have a broad spectrum of biological activity, especially as antifungal and antibacterial (Adeyemi et al., 2019; Butt et al., 2019; Hadi et al., 2019; Iqbal et al., 2015; Jamil et al., 2009; Javed et al., 2017). Biodiesel can be produced through several processes but transesterification, either by alkaline or acid catalysis, is typically used (Atadashi et al., 2013; Lam et al., 2010; Thangaraj et al., 2019). It is noteworthy that the class of organotins bearing four substituents groups, i.e., organotin (IV) compounds, are extensively consumed by timber, plastics, and paint industries in manufacturing

processes as PVC polymerization and stabilization, as well as building of heat-insulating and waterproof materials., catalysis of rubber, polyimide foams, and polyurethane (PU) (Erofeev et al., 2016; Finnegan et al., 2018; Matveeva et al., 2018; Yousif et al., 2016). It has been shown that organotin (IV) compounds also can serve as excellent catalysts for biodiesel transesterification and esterification, with many advantages over traditional ones such as no need for neutralization, no residual effluent production, little residual concentration in the final product, or even its immobilization in a heterogeneous phase (Hadi et al., 2019; Meneghetti et al., 2013; Zubair et al., 2019).

Organotin (IV) compounds can be broadly categorized in four types: R_4Sn (type I); R_3SnX (type II); R_2SnX_2 (type III); $RSnX_3$ (type IV), R representing an organic substituent linked to the tin atom by a tin–carbon bond, and X standing for a substituent (organic or not). The antimicrobial activity of tin (IV) is reported in many organic associations (Adeyemi et al., 2019; Ashfaq et al., 2004; Butt et al., 2019; Hadi et al., 2019; Jamil et al., 2009; Javed et al., 2017). As it has been long shown, depending on its substituents, its antimicrobial activity will vary (Van der Kerk & Luijten, 1954). For instance, triorganotin compounds (bearing three C–Sn bonds) tend to be more toxic than its counterparts bearing two substituents (Appel, 2004; Barbosa et al., 2018; Hadi et al., 2019; Hoch, 2001; Song et al., 2006). Actually, antimicrobial activity seems to be directly proportional to R substituent, with little impact of X substituent (Baul, 2008; Doctor & Fox, 1982; Hadi et al., 2019; Pellerito et al., 2006).

Regarding its substituents, it has been observed that alkyl (saturated linear-derived) are more toxic than aryl (aromatic-derived) (Doctor & Fox, 1982; Hadi et al., 2019). It is also reported that the hydrophobicity of the entire molecule is directly related to toxicity (Adeyemi et al., 2018)—for a high antifungal activity the total number of carbon atoms in the alkyl groups of a trialkyl alkyltin compound should be about 9–12 for instance (Hadi et al., 2019). This hydrophobicity is related to the cell membrane permeability. Depending on the substituent present, it binds to CoA-binding pockets (active sites) causing different effects once inside the cell: it is possible that organotin molecules either bind to metabolic crucial proteins or to the double-stranded DNA in an intercalate form (Butt et al., 2019; Lu et al., 2017). In fact, energy metabolism impairment in abalones by tributyltin along with a series of secondary metabolites accumulation suggests impairment of several metabolic pathways (Javed et al., 2017; Lu et al.,

2017), while DNA binding by butyltin compounds is also related (Barbosa et al., 2018; Butt et al., 2019; Iftikhar et al., 2018), as well as histones (Osada et al., 2005). Dibutyltin compounds show excellent catalytic activity in biodiesel production (Mônica A. da Silva et al., 2019; Mônica Araújo da Silva et al., 2017; Meneghetti et al., 2013) and less hazardous effects than tributyltin (Boyer, 1989; Hoch, 2001; Iqbal et al., 2015). Considering the promising advantages of catalysis using diorganotin (IV) complexes are its efficiency, safety, many aspects related to antimicrobial properties of biodiesel produced in those conditions are still unknown. The aim of this work was to evaluate the antimicrobial activity of organometallic (IV) tin-based catalysts used in the production of biodiesel. Two catalysts were evaluated on an uncharacterized inoculum according to ASTM E1259-18 standard (ASTM, 2018) and the best catalyst was also tested on a filamentous fungus (*Pseudallescheria boydii*), a yeast-like fungus (*Aureobasidium pullulans*), a Gram-negative bacterium (*Pseudomonas aeruginosa*), and a Gram-positive bacterium (*Bacillus pumilus*) as well as on uncharacterized inoculum.

Materials and methods

Biodiesel samples production using tin (IV) catalysts Biodiesel samples were obtained using an methanol:oil:catalyst molar ratio of 400:100:1 at 150 °C, in a 100-mL stainless steel reactor with stirring and temperature control (Parr Instrument Company, 4590 series). After 2 h of reaction, the reaction mixture was washed three times with distilled water, dried in the presence of MgSO₄, and centrifuged. The catalysts used were dibutyl tin dilaurate (Bu₂Sn (Lau)₂, 1—CAS nr. 77-58-7), di-*n*-butyl-oxo-stannane (Bu₂SnO, 2—CAS nr. 818-08-6), and sodium methoxide (CAS nr 124-41-4) to comparison. The yield of the transesterification reaction was determined by HPLC and expressed in terms of the percentage of fatty acid methyl esters (%)

FAMEs) produced. The amount of Sn present in the samples was determined by ICP-OS. All biodiesel samples used for microbiological tests were sterilized using 0.22-µm membranes (Millipore, USA) in a kitassato system (sterilized as well) as soon as received and kept at 4 °C in glass bottles protected from light by aluminum foil for until one month for microbiological testing.

Microorganisms

Four deteriogenic biodiesel microorganisms were used: a filamentous fungus (*Pseudallescheria boydii*), a yeast-like fungus (*Aureobasidium pullulans*), a Gram-negative bacterium (*Pseudomonas aeruginosa*), and a Gram-positive bacterium (*Bacillus pumilus*). An uncharacterized inoculum was also tested according to ASTM E1259 standard practice (ASTM, 2018).

Fungi were identified by sequencing of the ITS region, *B. pumilus* by sequencing of the 18S region of the rRNA, and *P. aeruginosa*, a clinical isolate identified by MALDI-TOF. The final inoculum concentration was 10⁵ cells or spores mL⁻¹.

Preparation of inocula

The bacterial inocula (*B. pumilus* and *P. aeruginosa*) were prepared from culture in tryptic soy broth (TSB) after 24 h stirring (200 rpm) and incubation at 30 °C. The inocula of *P. boydii* and *A. pullulans* were prepared from 7 day cultivation cultures on malt agar in slanted tubes in incubator at 28 °C by adding 2 mL of sterile (0.85%) saline and 2 mL of Tween 80 liquid solution (0.01%). Uncharacterized inoculum was prepared according to ASTM E1259 (ASTM, 2018). Briefly, 2% of B100 mixture (previously filter-sterilized) was added to 100 mL Bushnell-Haas mineral medium in an Erlenmeyer flask, which was then inoculated with 5 mL of microbiological sludge from a B100 tank. The flask was incubated in an orbital shaker at 29 ± 1 °C, 200 rpm for 7 days. The final concentration of all inocula was 10⁵ cells or spores mL⁻¹.

Microcosm preparation

Microcosm triplicates were prepared in 20-mL glass vials containing 4 mL of the oil phase (biodiesel with or without DBTDL) and 6 mL of Bushnell-Haas (BH) mineral broth as the aqueous phase. Three concentrations of DBTDL (3000, 1500, and 750 ppm) were tested

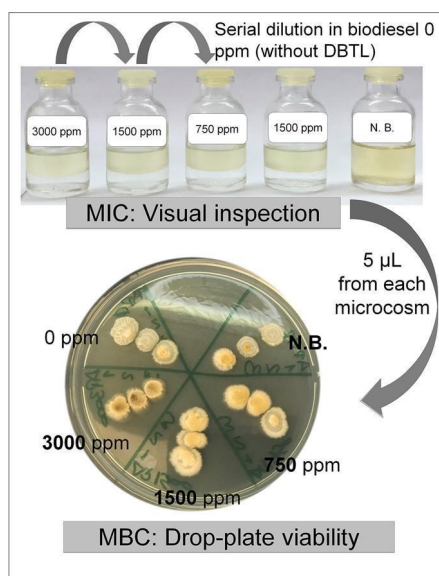


Fig. 1 Procedure used for the assembly and evaluation of the inhibitory and biocidal concentration of two tin-based catalysts used for biodiesel production with microorganisms during 14 days for bacteria and 28 days for fungi

and compared with biodiesel produced without the tin-based catalyst (0 ppm). Also, viability controls were made using TSB (bacteria) or Sabouraud broth (fungi and ASTM inoculum). All flasks were kept at 30 °C for 14 days (bacteria) or 28 days (fungi and ASTM inoculum) and analyzed by visual inspection and drop plate viability tests after 24, 48, and 72 h; 5, 7, and 14 days (bacteria); 21 and 28 days (fungi and ASTM inoculum). Drop plate tests were performed to evaluate MIC, inoculating 5 µL of the microcosm aqueous phase in agar (Sabouraud for fungi, TSA for bacteria), and visual inspection of biomass production was made to evaluate MBC. Final biomass production was defined by cell-forming unit (CFU) count (bacteria) or gravimetric

Fig. 2 Visual aspect on HAZE scale according to ASTM D4176 (04-2019) (ASTM International. American Society for Testing and Materials., 2019): a Biodiesel without tin and b biodiesel produced with $Bu_2Sn(Lau)_2$ containing 3000 ppm of tin

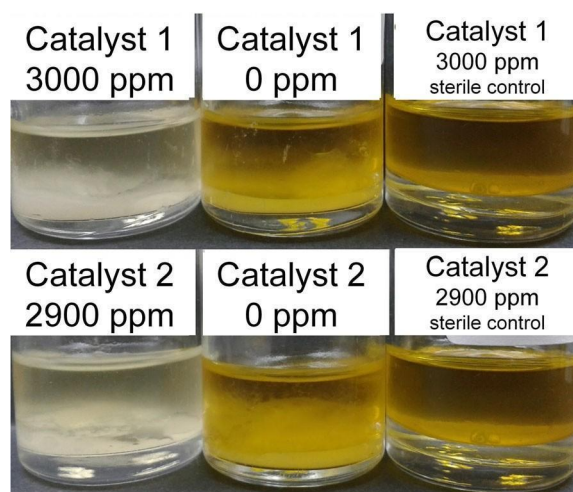
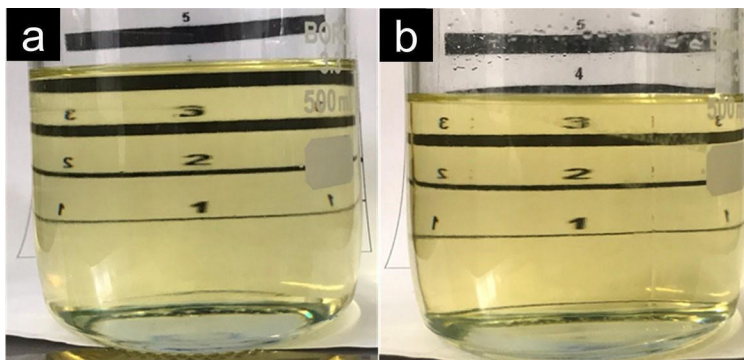


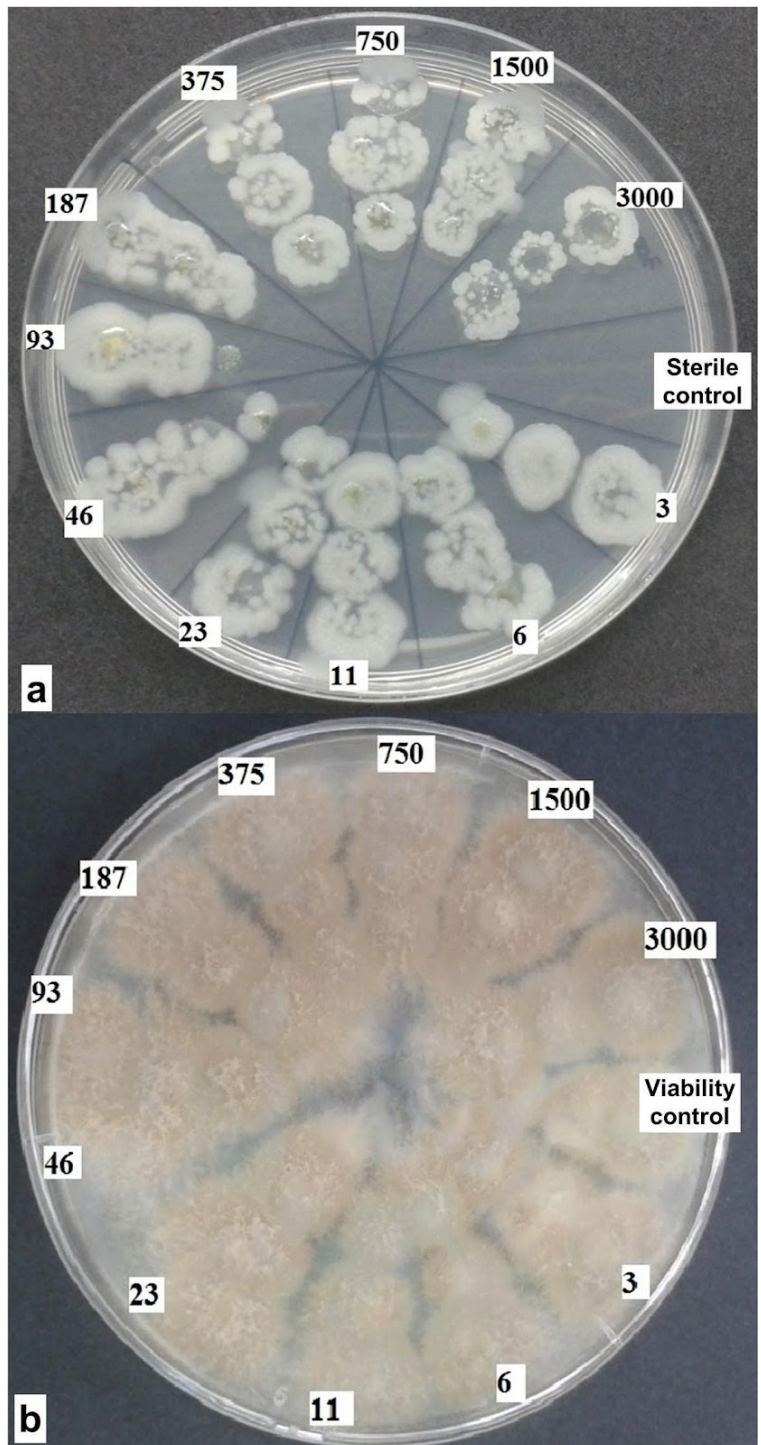
Fig. 3 Microcosm appearance of biodiesel produced with catalyst 1 (3000 ppm), catalyst 2 (2900 ppm), and without (0 ppm) tin-based catalyst in the presence of ASTM E1259 inoculum after 14 days. Sterile control contains no inocula

method after a 14-µm membrane filtration (fungi and uncharacterized inoculum).

CFU count—bacteria

After 14 days, an aliquot of the aqueous phase was serially diluted from 10^{-1} to 10^{-9} and 5 µL of each sample was deposited in triplicates of plate count agar plates. This drop plate method allows one to infer the final cell count of each bacterial microcosm similarly to a spread plate technique (Herigstad et al., 2001). Triplicate values were averaged and different dilutions were corrected accordingly. These procedures are summarized in Fig. 1.

Fig. 4 Viability tests through drop plate technique in plate count agar (a) and potato dextrose agar (b). Numbers indicate concentration of organotin catalyst. Viability control contains no organotin compound



ppm →	3000	1500	750	375	187	93	46	23	11	6	0
Catalyst 1	31.3	37.1	29.7	26.4	35.4	26.8	26.6	22.3	30.9	23.9	21.9
Catalyst 2	38.1	31.2	25.1	31.2	29.6	27.5	27.8	22.8	26.3	29.4	27.7

Table 1 Biomass amount for each concentration tested of catalysts 1 and 2; (ppm), parts per million of each catalyst tested Biomass amount (mg)

Biomass weight—fungi and ASTM inocula

After 28 days, the biomass formed was filtered through previously weighed filter paper. To remove adhered fuel from the biomass, discs were filter-washed with 3 ml of hexane. They were then placed at 30 °C for 48 h and transferred to a dehydrating chamber for 24 h to remove water, and the dry weight was recorded. Biomass weight was calculated as final

weight minus initial weight (in milligrams); triplicate values were averaged.

Chromatographic analysis

The fatty acid compositions of the samples were determined by gas chromatography (GC) using the AOCS official methods Ce 1-62 and Ce 2-26 (AOCS, 1998) and, the instrumental precision of the method was evaluated through 5 injections of a sample, with the determination of the standard deviation.

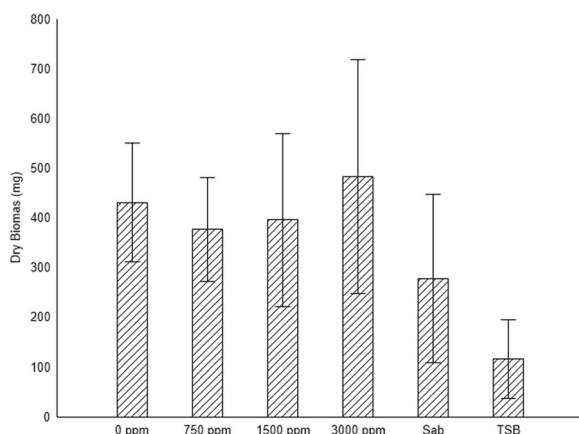


Fig. 5 Biomass values for ASTM inoculum in the presence (3000, 1500, and 750 ppm) and without tin (0 ppm) and viability control (SAB, Sabouraud nutrient broth; TSB, tryptic soy broth) after 28 days. Upper lines indicate standard error ($p = 0.05$)

Infrared analysis

Samples of the oil phase from initial and final times were analyzed by infrared spectroscopy with Fourier transform coupled to an attenuated total reflectance accessory (FTIR/ATR) in the Nicolet IR200 spectrometer (Thermo Fisher, USA) with a zinc selenide crystal (ZnSe). The spectra were obtained in the range of 4000–900 cm^{-1} , with a resolution of 4 cm^{-1} and 32 scans at 25 °C.

ICP-OES characterization

The amount of tin was determined by inductively coupled plasma optical emission spectrometry (ICP)

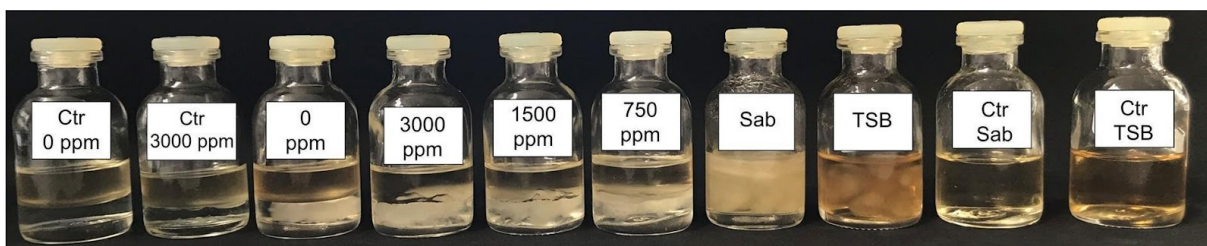


Fig. 6 Microcosm appearance of biodiesel produced with (3000, 1500, and 750 ppm) and without (0 ppm) tin-based catalyst in the presence of ASTM E1259 inoculum after 28

days. Sab, Sabouraud nutrient broth viability control. TSB, tryptic soy broth viability control. Ctr, Sterile controls

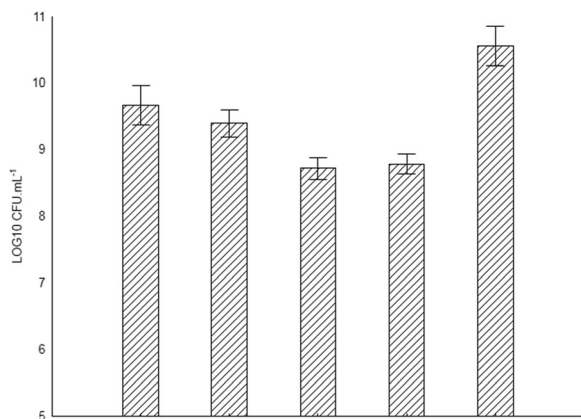


Fig. 7 CFU number of *P. aeruginosa* in the presence of tin (3000, 1500, and 750 ppm), without tin (0 ppm) and in the positive control (TSB) after 28 days. Upper lines indicate standard error ($p = 0.05$). X-axis initial cell inoculum of 10^5 cells mL^{-1}

OES) using Spectro Arcos equipment, after acid diges- tion of the samples.

Statistics

Statistical analyses of HRGC peaks percentage, pH, and biomass values (analysis of variance; Tukey test, $p = 5\%$; biomass values used p corrected by Sidak), were performed with PAST (version 3.25). Microbial growth and pH graphics were built using Statistica 10 (Statsoft Inc., 2011).

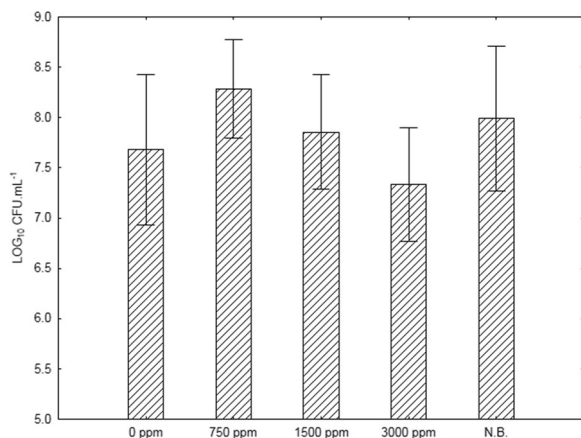


Fig. 8 CFU number of *P. aeruginosa* in the presence of tin (3000, 1500, and 750 ppm), without tin (0 ppm) and in the positive control (N.B) after 28 days. Upper lines indicate standard error ($p = 0.05$). X- axis corresponds to initial cell inoculum of 10^5 cells mL

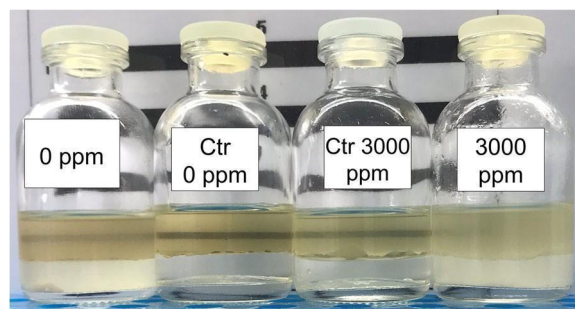


Fig. 9 Microcosm appearance of biodiesel produced with (3000 ppm) and without (0 ppm) tin-based catalyst, in the presence of *Pseudomonas aeruginosa* after 28 days. Ctr, sterile control. Lines behind flasks are from ASTM haze rating chart. Line number 2 is shown right behind the oil (upper) phase

Results and discussion

Biodiesel production and characterization

Biodiesel samples were obtained with $> 96\%$ ester con- tent (as determined by HPLC) using dibutyl tin dilaurate ($\text{Bu}_2\text{Sn}(\text{Lau})_2$, 1) and di-*n*-butyl-oxo-stannane (Bu_2SnO , 2) as catalysts and, the tin content, determined by ICP-OS, was 3000 ppm in both samples. A sample of biodiesel produced using sodium methoxide as catalyst was used as comparison. Biodiesel characterization is in accordance to ANP specifications, and the ester content of 97.8%, 98.0%, and 97.5% were found for biodiesel produced using 1, 2, and sodium methoxide as catalyst, respectively. Concerning the viscosity and density values of 4.4, 4.7, and 4.5 $\text{mm}^2 \text{s}^{-1}$ and 887, 886, and

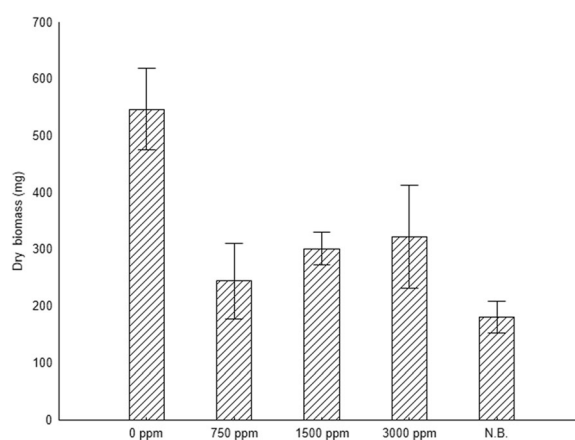


Fig. 10 Biomass values for *Aureobasidium pullulans* in the pres- ence (3000, 1500, and 750 ppm), and without tin (0 ppm) and viability control (N.B.—Sabouraud nutrient broth) after 28 days. Upper lines indicate standard error ($p = 0.05$)

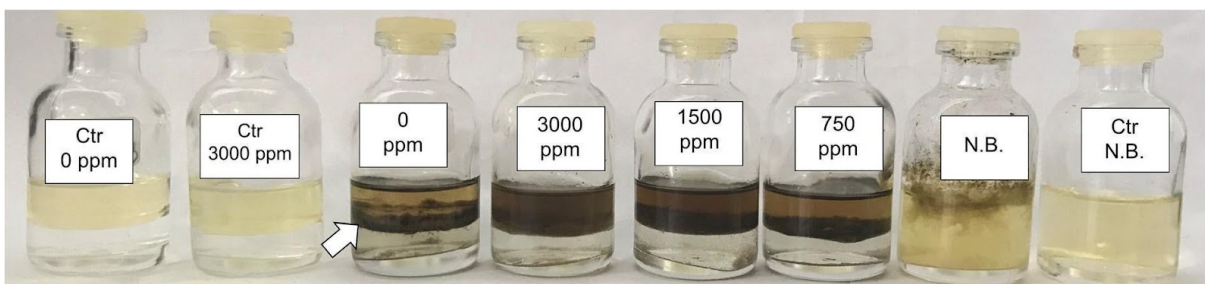


Fig. 11 Microcosm appearance of biodiesel produced with (3000, 1500 and 750 ppm) and without (0 ppm) tin-based catalyst in the presence of *A. pullulans* after 28 days. Arrow, melanin production. N.B., Sabouraud nutrient broth viability control. Ctr, sterile control vials

885 kg/m⁻³ were determined, respectively, for biodiesel produced using 1, 2, and sodium methoxide catalysts.

The visual aspect of control and the biofuels used in this experiment are in according to ASTM D4176 (04- 2019) HAZE scale (ASTM, 2019) for biodiesel produced using sodium methoxide and Bu₂Sn (Lau)₂ (Fig. 2).

Uncharacterized inoculum

Initially, the two catalysts (1 and 2) were tested only against the ASTM inoculum, exhibiting no inhibitory effect as evidenced by visual inspection (Figure 3) as well as viability tests (drop plate—Fig. 4) and biomass values (Table 1).

In the second part of the study, only biodiesel produced by 1 (Bu₂Sn (Lau)₂) and sodium methoxide catalysts were used. Similar results to the previous experiment were observed: visible biomass formation was observed in all ASTM E1259 inocula (Fig. 5) as it can be confirmed by dry biomass results (Fig. 6).

Bacteria

After 14 days, the bacterium *B. pumilus* showed inhibition proportional to the tin concentration (Fig. 7).

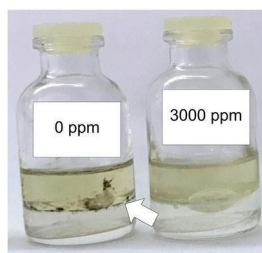


Fig. 12 Detail of microcosm appearance of biodiesel produced without (0 ppm) and with 3000 ppm tin-based catalyst in the presence of *A. pullulans* after 7 days. Arrow, biomass production

Conversely, *P. aeruginosa* was able to grow in all catalyst 1 concentrations tested (Fig. 8), exhibiting as well alterations in the texture of the aqueous phase (Fig. 9).

Fungi

The growth of the yeast-like fungus *A. pullulans* was also inhibited, with high residual tin concentrations, and melanin production by the fungus was observed in the microcosms (dark color—Fig. 10).

Melanin production began after 7 days of growth, which is indicated by a soon color change in biodiesel (Fig. 11).

Biomass production after 7 days also seemed to be lower in biodiesel with residual organotin compound (Fig. 12), in agreement with the final result after 28 days (Fig. 11).

The filamentous fungus *P. boydii* exhibits no difference in final dry biomass results (Fig. 13) among

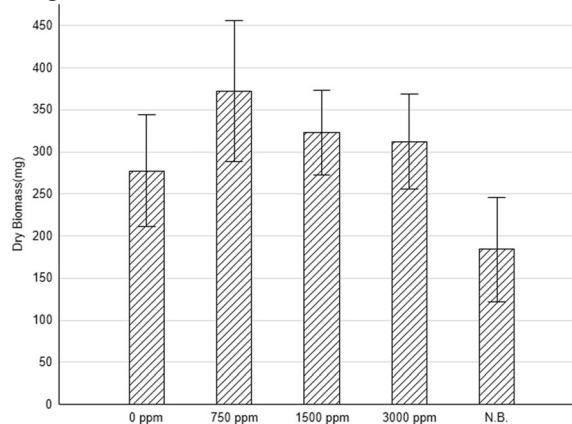


Fig. 13 Biomass values for *Pseudallescheria boydii* in the presence (3000, 1500, and 750 ppm), and without tin (0 ppm) and viability control (N.B.—Sabouraud nutrient broth) after 28 days. Upper lines indicate standard error ($p = 0.05$)

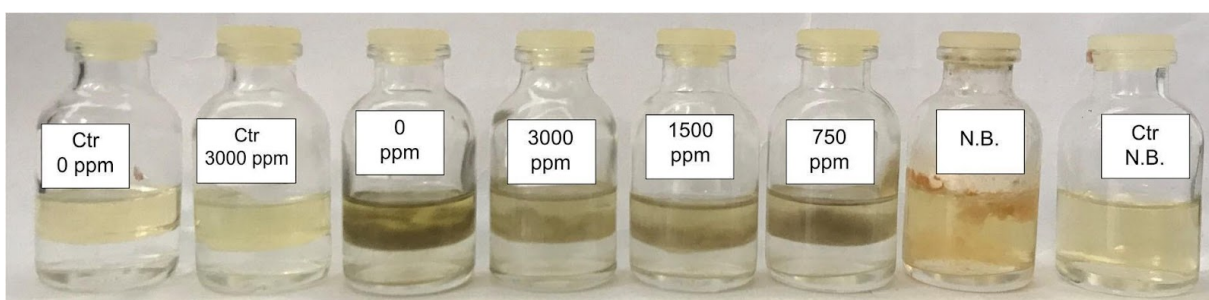


Fig. 14 Microcosm appearance of biodiesel produced without (0 ppm) and with (3000, 1500, and 750 ppm) tin-based catalyst in the presence of *Pseudallescheria boydii* inoculum after 28 days. N.B., Sabouraud nutrient broth viability control. Ctr, sterile controls

catalyst 1 concentrations, exhibiting visible biomass formation too (Fig. 14).

Despite this fact, visual inspection of microcosms show pronounced differences between *P. boydii* in 0 and 3000 ppm DBTD (Fig. 15).

As observed in Fig. 16, all experiments show acidification in the aqueous phase.

Chromatographic and infrared analysis

After 28 days of inoculum, oil phase samples were characterized by infrared spectroscopy and by gas chromatography. Table 2 shows the fatty acids composition of the oil phase, determined by gas chromatography.

The same samples were characterized by medium infrared spectroscopy. Figure 17 presents the spectra for biodiesel as received, with 0 ppm (a) and 3000 ppm (b) of residual organotin.

In Fig. 18, it is shown infrared spectra for sterile control, which is, biodiesel in contact with BH medium after 28 days of storage. Both products (as received and sterile control) exhibit no difference among spectra.

On the other hand, the infrared spectrum of ASTM oil phase shows a different result, as shown in Fig. 19.

The ASTM E1259 inoculum was able to grow in all concentrations of the tin-based residual catalyst. It is reported that uncharacterized inocula such as these are capable of growth in biodiesel, as well as biodiesel degradation (Bücker et al., 2014, 2018)

B. pumilus showed inhibition values suggesting a biocidal and sporistatic effect, as reported in the literature (Butt et al., 2019; Javed et al., 2017).

P. aeruginosa ability of growing in biodiesel is well reported (Bücker et al., 2014; De Azambuja et al., 2017; Jensen et al., 2017). Several strains of this species also produce a common emulsifier called Rhamnolipid (Cieśła et al., 2018; Fenibo et al., 2019; Kaskatepe & Yildiz, 2016). This substance has several applications, including environmental remediation of metal ions and organometallic compounds such as organotin complexes (Cieśła et al., 2018; Finnegan et al., 2018). As hydrophobicity and molecule partition in the aqueous phase are crucial to organotin (IV) toxicity (Adeyemi et al., 2018), this emulsifier may have led to a reduced antimicrobial effect in aqueous medium keeping it from entering the cell.

It is known that melanin production may have a variety of functions, from virulence factor to stress

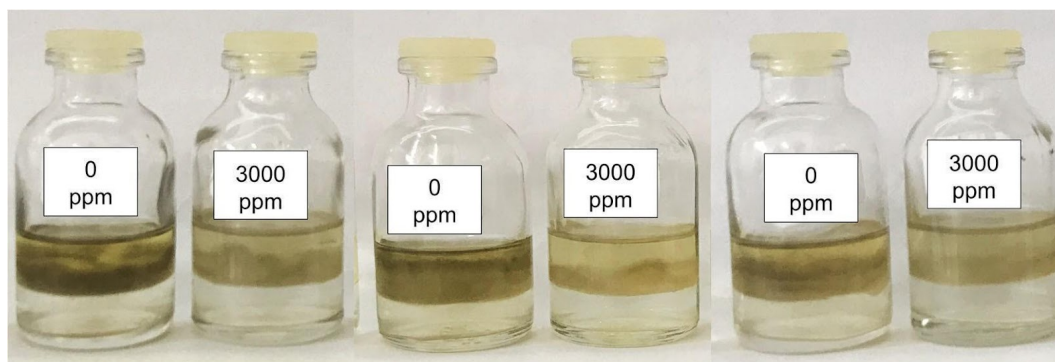


Fig. 15 Microcosms triplicate of *P. boydii* after 28 days of incubation

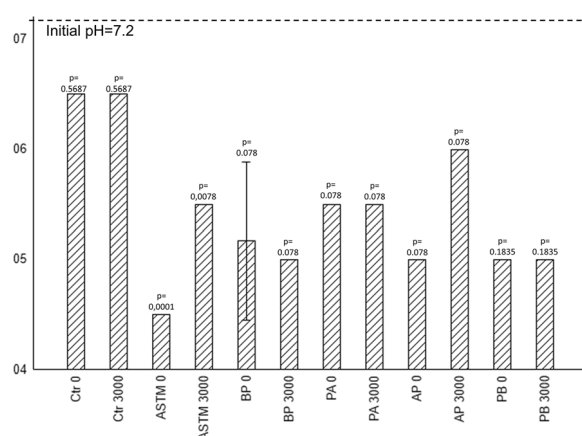


Fig. 16 Graphic showing final pH of each inoculum compared to sterile controls. Upper dotted lines indicate initial pH value (7.2). Spreadlines indicate standard error ($p = 0.05$). P values are relative to initial pH

response (Cordero & Casadevall, 2017; Wang et al., 2019). The fact that more melanin production by

A. pullulans is observed, in the absence of organotin, may indicate inhibition of its production, or inhibition of growth of the yeast itself. It is probable that both scenarios would take place due to alterations in DNA expression, since it is reported that dibutyltin compounds can intercalate the DNA and bind histones (Javed et al., 2017; Osada et al., 2005).

Table 2 Fatty acid methyl esters (FAMES) composition determined by gas chromatography. 0 ppm indicates biodiesel without residual organotin. 3000 ppm indicates biodiesel with 3000 ppm of residual organotin. AR indicates sample was analyzed as received. Ctr indicates control samples that were in contact with BH

Samples	Palmitic acid	Stearic acid	Oleic acid	Linoleic acid	Linolenic acid
0 ppm AR	11.7	3.7	22.5	55.6	6.5
3000 ppm AR	12.5	4	23.3	55.3	4.9
0 ppm Ctr1	12.3	3.8	22.5	55.9	5.6
3000 ppm Ctr1	12.5	3.9	22.9	55.6	5.2
0 ppm Ctr2	12.8	4	23.4	54	5.9
3000 ppm Ctr2	12.9	3.9	23.3	54.1	5.8
0 ppm ASTM	13.4	3.7	22.3	54.5	6.2

It is known that *P. boydii* is an emergent pathogen, with antimicrobial resistance reported (Coates et al., 2020; Gilgado et al., 2006; Liu et al., 2019; Livermore, 2002; Yagüe et al., 2013). In spite of the incipient use of organotin substances for microbial control, it is also reported that this substance already showed a medium effect over environmental isolates of *P. boydii* (Liu et al., 2019), which could explain its resistance.

pH medium alterations may indicate production of either secondary acid metabolites or hydrolysis products (Fátima M. Bento & Gaylarde, 2001; Bückner et al., 2011, 2014; Cazarolli et al., 2020)

The final pH observed for both sterile controls (0 and 3000 ppm) was 6.5, while for *B. pumilus* it was 6.0 and for *P. aeruginosa* was 5.5. Both fungi exhibited a final pH of 5.0 in both catalyst 1 concentrations. Since initial pH of aqueous phase was 7.2, the acidification observed in the sterile control was probably due the tendency of migration of FAMES from the oil phase and hydrolysis products to the aqueous phase (Fátima

M. Bento & Gaylarde, 2001; Bückner et al., 2011, 2014; Cazarolli et al., 2020).

Previous studies show consistently that, despite the visible microbial growth evidenced by biomass formation and dry weight data, GC analysis of FAMES after incubation seemed unaltered. It has been reported (Cazarolli et al., 2014) that several methyl esters migrate

medium Ctr 1: sterile control for isolated microorganisms experiment; Ctr2: sterile for ASTM inoculum. ASTM, ASTM inoculum; BP, *Bacillus pumilus* inoculum experiment; AP, *Aureobasidium pullulans* inoculum experiment. Standard deviation is $\pm 1.0\%$

Samples	Palmitic acid	Stearic acid	Oleic acid	Linoleic acid	Linolenic acid
0 ppm AR	11.7	3.7	22.5	55.6	6.5
3000 ppm AR	12.5	4	23.3	55.3	4.9
0 ppm Ctr1	12.3	3.8	22.5	55.9	5.6
3000 ppm Ctr1	12.5	3.9	22.9	55.6	5.2
0 ppm Ctr2	12.8	4	23.4	54	5.9
3000 ppm ASTM	13.4	3.8	22.6	54.2	6.1
0 ppm BP	13	3.6	22.6	54.7	6.1
3000 ppm BP	13.2	3.8	22.8	54.1	6.1
0 ppm AP	12.8	4.4	23.6	53.4	5.8
3000 ppm AP	12.8	3.6	22	55.1	6.4

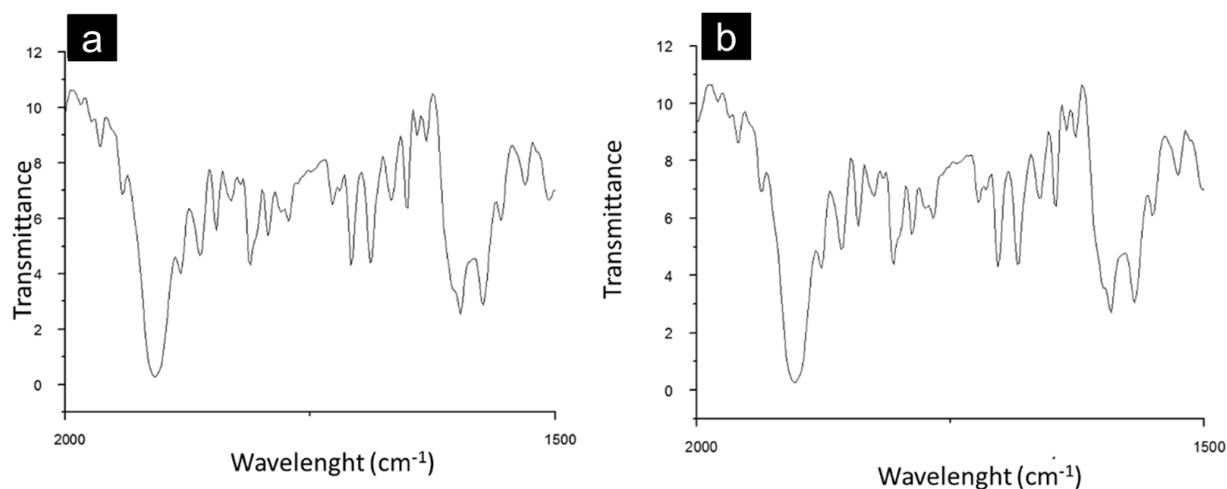


Fig. 17 Infrared spectra between 2000 and 1500 cm^{-1} of biodiesel as received, with 0 ppm (A) and 3000 ppm (B) of residual organotin

from the oil phase to the aqueous phase, thus providing a carbon source for microbial growth, and yet maintaining the FAMES composition of the oil phase roughly the same. These findings also help to explain the spontaneous pH reduction observed in the sterile controls.

Beker et al. (Beker et al., 2016) obtained similar results regarding FAMES composition after microbial incubation, suggesting that this migration may have contributed to the microbial population nutrients for its growth, along with other biodiesel components such as glycerin, mono, di and triacylglycerides compounds (Beker et al., 2016; Bucker et al., 2011; Cazarolli et al., 2014).

The alteration around 1710 cm^{-1} observed in

Fig. 19 (a and b) can be related to biodiesel degradation (Boelter et al., 2018; Pinho et al., 2014). This is an important result

as biomass observations (as shown in Fig. 6) also indicate inoculum growth. According to GC analysis, there is no modification in the fatty acid profile after 28 days and the changes in the carbonyl region observed in FTIR spectra may be related to acid formation, due to hydrolysis of TAGs and degradation reactions. These chemical species tend to migrate to the aqueous phase, causing the aforementioned pH reduction in the sterile controls (Beker et al., 2016; Bucker et al., 2011; Cazarolli et

al., 2014).

These findings allow studies in larger scales to be outlined, i.e., field and industrial scales, conducted using safe and useful concentrations of the organotin residual concentrations after a scaling-up production with industrial purposes. Also, environmental models can be employed to ensure its safety such as plant and animal models in contact with residual bottom water.

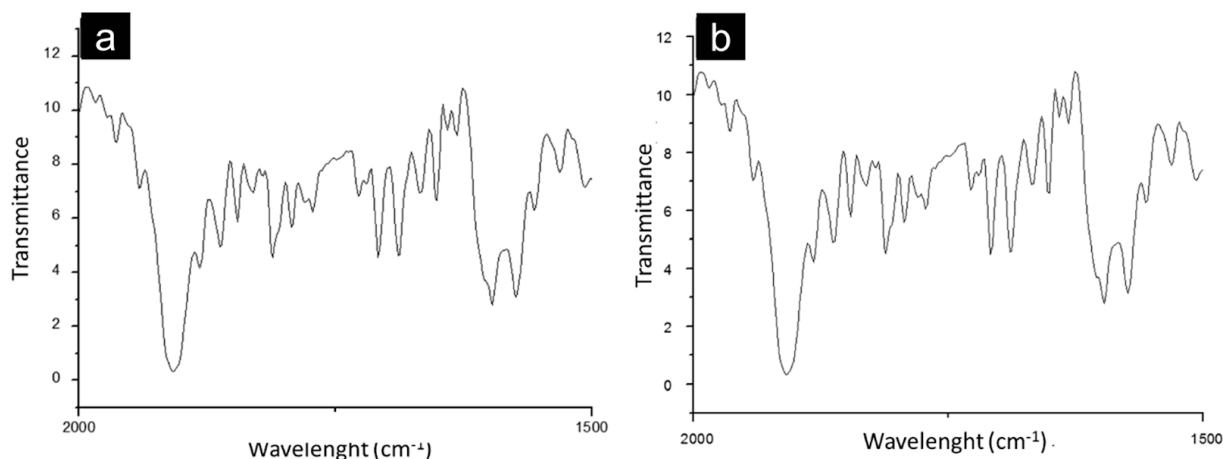


Fig. 18 Infrared spectra between 2000 and 1500 cm^{-1} of biodiesel with 0 ppm (A) and 3000 ppm (B) of residual organotin after 28 days in contact with BH medium

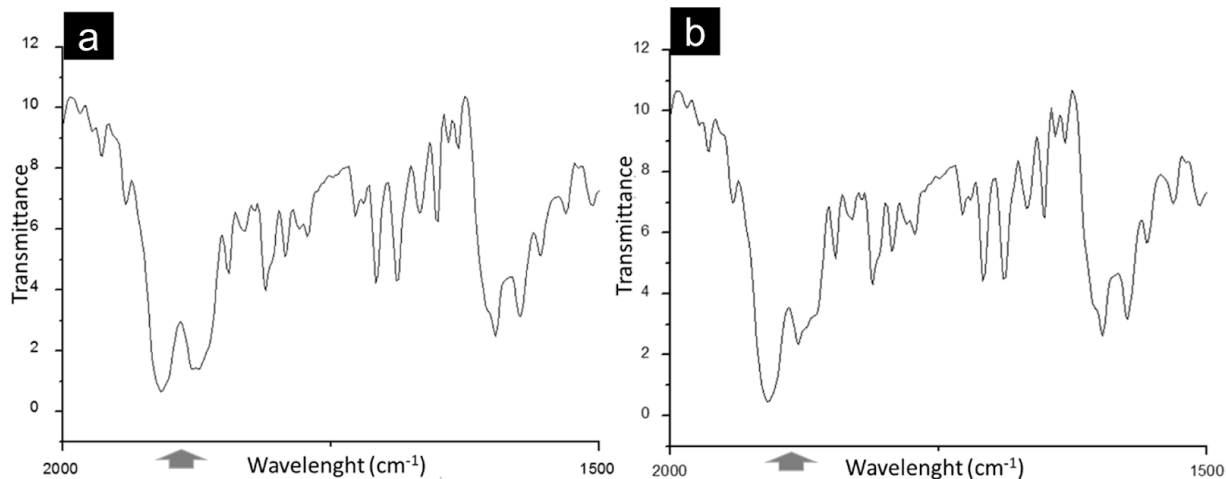


Fig. 19 Infrared spectra between 2000 and 1500 cm^{-1} of biodiesel with 0 ppm (A) and 3000 ppm (B) of residual organotin after 28 days in contact with ASTM inoculum. Arrow, region of wavelength related to carbonyl groups

Conclusions

The organotin compounds tested in the present work did not show any biocide effect over the microorganisms tested, an important novelty. Our findings in terms of microbial growth are in accordance to previous reports in the literature regarding the uncharacterized inoculum ASTM (Bucker et al., 2014, 2018) and the isolated organisms *P. boydii* (Boelter et al., 2018; Cazarolli et al., 2014; Martin-Sanchez et al., 2018), *A. pullulans* (Cazarolli et al., 2020), *B. pumilus* (Beker et al., 2016), and

P.aeruginosa (Bucker et al., 2014; De Azambuja et al., 2017; Jensen et al., 2017). These results also suggest that the tin residual content did not present environmental concern regarding this catalyst, as it had low or no effect at all over the microorganisms tested.

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7. DISCUSSÃO GERAL

Dois compostos organoestânicos (iv) presentes como resíduo da catálise de biodiesel, foram testados em inóculos ASTM e um deles foi também testado em espécies deteriorogênicas isoladas, mostrando pouca ou nenhuma atividade sobre os microrganismos, em concentrações próximas às permitidas em lei para aditivos.

Segundo relatos da literatura, aditivos baseados em estanho com efeito antimicrobiano e incrustante, especialmente aqueles associados a radicais orgânicos, são poluentes ambientais importantes, ocasionando efeitos adversos na fauna aquática (Formalewicz, 2019; Chen, 2017; Monteiro, 2021). Assim, seu uso já foi proibido por exemplo na tinta usada em cascos de navios, como agente anti-incrustante (U.E., 2003; Sonak, 2009), uma vez que há desprendimento lento destes compostos da tinta para o oceano (Lagerstörn, 2017). No entanto, o efeito antimicrobiano apresentado pelo composto de organoestanho aqui testado foi baixo, sugerindo três hipóteses: que seu uso seja seguro em termos ambientais, que os microrganismos testados possuam, todos, resistência intrínseca ou adquirida; ou que os compostos de organoestanho presentes em combustível não afetem os microrganismos da mesma forma que em meio aquoso.

Segundo Van der Kerk & Luijten, 1954 a toxicidade dos compostos organometálicos de forma geral, incluindo os de organoestanho em seus diferentes estados de oxidação (ii e iv, os mais comuns em moléculas organoestânicas) relaciona-se ao radical orgânico por ele apresentado, o que aumenta sua penetração em biomembranas (Baul, 2008; Doctor & Fox, 1982; Hadi et al., 2019; Pellerito et al., 2006) Neste caso o dibutilestanho apresenta menos toxicidade do que tributilestanho, tanto pelo fato de possuir 2 substituintes orgânicos em vez de 3, quanto pelo fato destes substituintes possuírem caráter menos hidrofóbico do que outros substituintes orgânicos com maior número de carbonos (Doctor & Fox, 1982; Hadi et al., 2019). Desta forma, compostos como dibutil-estanho possuem excelente atividade catalítica na produção de biodiesel (Mônica A. da Silva et al., 2019; Mônica Araújo da Silva et al., 2017; Meneghetti et al., 2013) e apresentam menos perigo do que compostos de tributilestanho (Boyer, 1989; Hoch, 2001; Iqbal et al., 2015).

Testes toxicológicos e ecotoxicológicos, bem como de emissões gasosas, ainda necessitam ser realizados nesse sentido para garantir a segurança ambiental de seu uso. Também seria preciso verificar se a presença de resíduos de estanho no combustível poderiam causar algum efeito indesejado em diferentes partes do motor, principalmente deposições metálicas nos cilindros de combustão e nos bicos injetores.

Da mesma forma, a presença de 1000 ppm de 8-hidroxiquinolina em combustível B20 com meio BH apresentou precipitações metálicas no fundo da fase aquosa. Embora esta porção não esteja em contato direto com o combustível, indica a interação de metais

presentes na fase aquosa com o composto presente na fase combustível - ou seja, um nível de partição suficiente para que fossem alcançadas as duas fases, um dos requisitos para que um antimicrobiano possa ter seu uso considerado em sistemas de combustível. O Clioquinol apresentou efeitos antimicrobianos semelhantes aos observados para 8-Hidroxiquinolina, molécula da qual este é derivado. Estes testes em microcosmos indicam o efeito destas moléculas, em comparação com o observado por antimicrobianos comerciais, em um cenário mais próximo daquele encontrado em tanques de estocagem. O efeito sinérgico observado para o antimicrobiano C, na presença de Clioquinol, prevê ao menos 50% de redução da quantidade necessária de morfolininas aplicadas no combustível para sua efetividade, com apenas 16µg/L de clioquinol - em comparação com 250 µg/L de clioquinol sem morfolininas, e 1mg/L da mistura de morfolininas.

Segundo a especificação do diesel brasileiro, seguindo uma tendência mundial, a quantidade máxima de enxofre é de 10 µg/L. Por outro lado, diversos biocidas aprovados pela EPA, como isotiazolonas, possuem enxofre em sua constituição, o que já descredencia seu uso atualmente em misturas Bx. Uma das vantagens do resultado observado neste trabalho é justamente a ausência de enxofre em todas as moléculas testadas - tanto a mistura de morfolininas (Metileno-bis-morfolina / nitro metil-morfolina) quanto 8-hidroxiquinolina e seu derivado, clioquinol - são isentos de enxofre. Além disso, seu uso na dose de 16 ppm (Log25 vezes menor) foi suficiente para diminuir à metade a quantidade de morfolininas necessárias para o mesmo efeito inibitório, indicando um efeito sinérgico de uma molécula isenta de enxofre, precipitações ou outras alterações de especificação observáveis.

Este trabalho traz duas contribuições importantes no sentido de elucidar o efeito antimicrobiano de compostos em dois cenários: sua presença residual após a catálise (dibutil-organoestanho) e sua adição para controle microbiológico (clioquinol, 8-hidroxiquinolina e compostos comerciais). Observou-se que a associação de compostos de classes diferentes, com mecanismos de ação distintos sobre microrganismos, é uma importante alternativa a ser utilizada. A associação de antifúngicos em aplicações clínicas já é uma prática usual atualmente, mas esta mesma prática pode trazer benefícios em cenários como o industrial e de combustíveis, reduzindo impactos ambientais e econômicos.

Também é importante ressaltar que, assim como a adição de compostos ao combustível pode ocasionar interações e alterações em suas especificações, a associação de compostos biocidas também precisa ser feita criteriosamente, de modo a evitar alterações nas propriedades do combustível, dos próprios biocidas, e efeitos antagônicos sobre os microrganismos deteriorogênicos. Em suma, a análise e escolha de compostos químicos para controle microbiológico em combustíveis deve ser feita de forma criteriosa, levando em conta as especificidades do cenário de armazenamento.

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